# The Molybdenotungsten Monophosphate $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ : An Original Three-Dimensional Framew ork Built Up of "MPO" Chains ( $\mathrm{M}=\mathrm{Mo}, \mathrm{W}$ ) 

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#### Abstract

A new molybdenotungsten monophosphate $\left(\mathrm{Mo}_{x} \mathrm{~W}_{2-x}\right) \mathrm{O}_{3}$ $\left(\mathrm{PO}_{4}\right)_{2}$ has been synthesized for $0 \leq x \leq 1.25$. The crystal structure of the phase $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ has been determined. This phase crystallizes in the $P 2_{1} / m$ space group with $a=7.827(1) \AA$, $b=12.538(1) \AA, c=7.833(1) \AA, \beta=92.36(1)^{\circ}$. The structure can be described from the assemblage of two kinds of $\left[M \mathrm{PO}_{8}\right]_{\infty}$ chains running along $b$, with a cis and trans orientation of the $\mathrm{PO}_{4}$ tetrahedra. Another description which consists of the stacking of two different sorts of $\left[M_{2} \mathrm{P}_{2} \mathrm{O}_{15}\right]_{\infty}$, layers along $b$ is also presented that involves two sorts of bioctahedral units " $M_{2} \mathrm{O}_{11}$ " with the perovskite and HTB configurations, respectively. © 1997 Academic Press


## INTRODUCTION

The association of molybdenum and tungsten in the same matrix of transition metal phosphates has allowed original mixed frameworks to be generated, different from the nonsubstituted molybdenum or tungsten phosphates. This has been shown for the phosphates $\mathrm{Na}_{x}(\mathrm{MoW})_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}(1)$, $\mathrm{Na}_{1+x}(\mathrm{MoW})_{2} \mathrm{O}_{5} \mathrm{PO}_{4}$ (2), $\mathrm{K}_{6} \mathrm{Mo}_{3} \mathrm{~W}_{9} \mathrm{PO}_{40} \cdot 13 \mathrm{H}_{2} \mathrm{O}$ (3), and $\mathrm{K}_{6.6} \mathrm{Mo}_{2.26} \mathrm{~W}_{3.74} \mathrm{P}_{4} \mathrm{O}_{31}$ (4). In this respect, it is of interest to study the substitution of molybdenum for tungsten in the tungsten phosphate $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ (5). Such an investigation is all the more interesting since only a structural model could be proposed for the latter tungsten phosphate.

The present paper deals with the study of the phase $\mathrm{Mo}_{x} \mathrm{~W}_{2-x} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$, with $0 \leq x \leq 1.25$, whose structure determination for $x=1$ shows that it exhibits an original three dimensional framework, different from the structural model proposed for $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$.

## CHEMICAL SYNTHESIS AND CRYSTAL GROWTH

The chemical synthesis of the phase $\mathrm{Mo}_{x} \mathrm{~W}_{2-x} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ was performed by heating stoichiometric mixtures of
$\mathrm{MoO}_{3}, \mathrm{WO}_{3}$, and $\mathrm{H}\left(\mathrm{NH}_{4}\right)_{2} \mathrm{PO}_{4}$ in air in a platinum crucible at 1070 K for 12 h . In these conditions a pure solid solution was obtained for $0 \leq x \leq 1.25$. Beyond $x=1.25$, the samples were melt and amorphous. Attempts to extend this solid solution by working at lower temperature led to a mixture of $\mathrm{WO}_{3}$ and $\left(\mathrm{MoO}_{2}\right)_{2} \mathrm{P}_{2} \mathrm{O}_{7}$ (6).

Two different methods were used for crystal growth of the phase $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$, that both involve a doping of the crystals with $\mathrm{Mo}(\mathrm{V})$.

In a first method, the stoichiometric mixture of $\mathrm{H}\left(\mathrm{NH}_{4}\right)_{2} \mathrm{PO}_{4}, \mathrm{WO}_{3}$, and $\mathrm{MoO}_{3}$ in the ratio 2:1:1 was heated at 600 K in a platinum crucible to decompose the ammonium phosphate. The resulting mixture was then heated at 873 K in an evacuated silica ampoule. A green powder with some green crystals on the silica ampoule were obtained. The microprobe analysis of the single crystals confirmed their cationic composition "MoWP 2 ," and the X-ray single crystal investigation evidenced a monoclinic cell similar to that observed for $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}(5)$.

The second method, starting from a mixture of nominal composition $\mathrm{Li}_{0.25} \mathrm{MoWP}_{2} \mathrm{O}_{11}$, allowed larger crystals of good quality to be grown. In a first step, the adequate mixture of $\mathrm{H}\left(\mathrm{NH}_{4}\right)_{2} \mathrm{PO}_{4}, \mathrm{Li}_{2} \mathrm{CO}_{3}, \mathrm{MoO}_{3}$, and $\mathrm{WO}_{3}$ in the ratio 2:0.125:0.96:1 was heated similarly at 600 K in a platinum crucible in order to decompose the ammonium phosphate and the carbonate. Then the resulting mixture was added to the required amount of molybdenum metal ( 0.04 mole), sealed in a evacuated silica ampoule, heated for 100 h at 873 K , cooled at $1 \mathrm{~K} / \mathrm{h}$ down to 773 K , and finally quenched at room temperature. Green, plate-like crystals mixed with a brown powder were obtained. The microprobe analysis of these crystals, as well as their crystal study, confirmed their cation composition " $\mathrm{MoWP}_{2}$ " and their monoclinic cell similar to that obtained by the first method.

The powder X-ray pattern of the $x=1$ composition of the solid solution synthesized at 1070 K could be perfectly indexed in the monoclinic cell deduced from the single crystal X-ray study (Table 1).

TABLE 1
Summary of Crystal Data, Intensity Measurements, and Structure Refinement Parameters for $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$

| Crystal data |  |
| :---: | :---: |
| Space group | $P 2_{1} / m$ |
| Cell dimensions | $\begin{aligned} & a=7.827(1) \AA, b=12.538(1) \AA, c=7.833(1) \AA \\ & \alpha=90.0^{\circ}, \beta=92.36(1)^{\circ}, \gamma=90.0^{\circ} \end{aligned}$ |
| Volume | $768.1(1) \AA^{3}$ |
| Z | 2 |
| Intensity measurements |  |
| $\lambda(\mathrm{MoK} \alpha)$ | 0.71073 A |
| Scan mode | $\omega-\theta$ |
| Scan width ( ${ }^{\circ}$ ) | $1.15+0.35 \tan \theta$ |
| Slit aperture (mm) | $1.15+\tan \theta$ |
| $\operatorname{Max} \theta\left({ }^{\circ}\right)$ | 45 |
| Standard reflections | 3 every h |
| Reflections measured | 6812 |
| Reflections with $I>3 \sigma$ | 1935 |
| $\mu\left(\mathrm{mm}^{-1}\right)$ | 17.1 |
| Structure solution and refinement |  |
| Parameters refined | 157 |
| Agreement factors | $R=0.027 \quad R_{\text {w }}=0.025$ |
| Weighting scheme | $w=1 / \sigma^{2}$ |
| $\Delta / \sigma$ max | $<0.01$ |

## STRUCTURE DETERMINATION

A green plate-like crystal with $0.130 \times 0.058 \times 0.013 \mathrm{~mm}$ dimensions was selected from the " $\mathrm{Li}_{0.25} \mathrm{MoWP}_{2} \mathrm{O}_{11}$ " mixture for the structure determination. The cell parameters reported in Table 1 were determined and refined by diffractometric technique at 294 K with least squares refinement based upon 25 reflections in the range $18^{\circ}<\theta<25^{\circ}$. They are similar to those obtained by Kierkegaard (5) for $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$. The systematic absences $k=2 n+1$ for $0 k 0$ are consistent with the space groups $P 2_{1}$ and $P 2_{1} / m$. The refinement of the structure was successful in the centrosymetric group $P 2_{1} / m$. As seen before by Kierkegaard for $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$, the $k$ odd reflections are very weak in whole space, so that a pseudo-translation $01 / 20$ should be involved. The structure was solved by the heavy atom method. The Mo and W atoms were first distributed at random in the three metallic $M$ sites. The refinement of the occupancy of the sites led to a preferential occupation of $M(1)$ by tungsten ( $55 \% \mathrm{~W}, 45 \% \mathrm{Mo}$ ), whereas the two other sites were preferentially occupied by molybdenum with $47 \%$ W and $53 \%$ Mo for $M(2)$ and $35 \% \mathrm{~W}$ and $65 \% \mathrm{Mo}$ for $M(3)$ (Table 2). Such a distribution of the metallic atoms leads to the formulation $\mathrm{Mo}_{1.04} \mathrm{~W}_{0.96} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ which is in perfect agreement with the microprobe analysis of the crystals. The oxygen and phosphorus sites were deduced from Fourier differences series. The refinement of all the atomic parameters and anisotropic thermal factors with a full matrix least squares method led to $R=0.027$ and

TABLE 2
Positional Parameters and Their Estimated Standard Deviations for $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$

| Atom | $x$ | $y$ | $z$ | $B\left(\AA^{2}\right)$ |
| :--- | :---: | :--- | :--- | :--- |
| $M(1)$ | $0.14929(4)$ | $0.49729(8)$ | $0.69218(4)$ | $0.78(1)$ |
| $M(2)$ | $0.3591(1)$ | 0.25 | $0.2021(1)$ | $0.57(3)$ |
| $M(3)$ | $0.3608(1)$ | 0.75 | $0.1898(1)$ | $0.68(3)$ |
| $\mathrm{P}(1)$ | $0.2526(2)$ | $0.4946(3)$ | $0.1173(2)$ | $0.62(3)$ |
| $\mathrm{P}(2)$ | $0.2494(3)$ | 0.75 | $0.6037(3)$ | $0.50(5)$ |
| $\mathrm{P}(3)$ | $0.0741(3)$ | 0.25 | $0.8468(4)$ | $0.70(5)$ |
| $\mathrm{O}(1)$ | $0.3309(6)$ | $0.4761(5)$ | $0.5881(7)$ | $1.7(2)$ |
| $\mathrm{O}(2)$ | 0.0 | 0.5 | 0.5 | $1.1(2)$ |
| $\mathrm{O}(3)$ | $0.1694(7)$ | $0.6534(5)$ | $0.6903(8)$ | $1.4(2)$ |
| $\mathrm{O}(4)$ | $0.2622(5)$ | $0.5050(8)$ | $0.9228(6)$ | $1.1(1)$ |
| $\mathrm{O}(5)$ | $0.1016(7)$ | $0.3473(5)$ | $0.7355(8)$ | $1.2(1)$ |
| $\mathrm{O}(6)$ | $-0.0767(6)$ | $0.5311(5)$ | $0.8295(7)$ | $1.4(2)$ |
| $\mathrm{O}(7)$ | $0.2080(9)$ | 0.25 | $-0.005(1)$ | $1.4(3)$ |
| $\mathrm{O}(8)$ | $0.3761(6)$ | $0.4034(5)$ | $0.1754(8)$ | $0.8(1)$ |
| $\mathrm{O}(9)$ | $0.204(1)$ | 0.25 | $0.346(1)$ | $1.4(2)$ |
| $\mathrm{O}(10)$ | $0.5662(9)$ | 0.25 | $0.017(1)$ | $1.0(2)$ |
| $\mathrm{O}(11)$ | $0.555(1)$ | 0.25 | $0.356(1)$ | $1.8(3)$ |
| $\mathrm{O}(12)$ | $0.1011(9)$ | 0.75 | $0.077(1)$ | $1.4(2)$ |
| $\mathrm{O}(13)$ | $0.3254(8)$ | $0.5949(5)$ | $0.2039(8)$ | $1.3(2)$ |
| $\mathrm{O}(14)$ | $0.220(1)$ | 0.75 | $0.413(1)$ | $1.0(2)$ |
| $\mathrm{O}(15)$ | $0.548(1)$ | 0.75 | $0.308(1)$ | $1.4(2)$ |

Note. Anisotropically refined atoms are given in the form of the isotropic equivalent displacement parameter defined as $B=4 / 3 \sum_{i} \sum_{j} \mathbf{a}_{i} \cdot \mathbf{a}_{j} \beta_{i j}$. $M(1)=\mathrm{W}_{0.55} \mathrm{Mo}_{0.45}, M(2)=\mathrm{W}_{0.47} \mathrm{Mo}_{0.53}, M(3)=\mathrm{W}_{0.35} \mathrm{Mo}_{0.65}$.
$R_{\mathrm{w}}=0.025$ and to the parameters in Table 2. One should emphasize that the (Mo, W) and oxygen sites, and half of the phosphorus sites are related by a pseudo-translation $01 / 20$; only the remaining phosphorus atoms destroy this pseudotranslation. This fact is consistent with the dramatic weakening of the $k$ odd reflections.

## DESCRIPTION OF THE STRUCTURE AND DISCUSSION

These results confirm the atomic positions determined for tungsten and three quarters of phosphorus by Kierkegaard (5) for $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$, and show that the oxygen and the last quarter of phosphorus sites are different from the structural model proposed by this author. Such a difference could be expected since the latter model did not take into account the $k$ odd reflections, leading to a $b$ parameter of $6.72 \AA$ instead of $12.54 \AA$.

The projection of the structure of this phase along $\mathbf{b}$ (Fig. 1) shows that the $\left[M_{2} \mathrm{P}_{2} \mathrm{O}_{11}\right]_{\infty}$ framework consists of bioctahedral units $M_{2} \mathrm{O}_{11}$ connected through single $\mathrm{PO}_{4}$ tetrahedra. Each octahedral unit is formed of two cornershared $\mathrm{Mo}_{6}$ octahedra, but two kinds of $M_{2} \mathrm{O}_{11}$ units are observed: in the first, corresponding to the $M(1)$ sites, the two octahedra form $\mathrm{O}-\mathrm{O}-\mathrm{O}$ angles of $90^{\circ}$ as in perovskite, whereas in the second, the two octahedra $M(2)$ and $M(3)$


FIG. 1. The projection of the structure of $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ along $\mathbf{b}$.
form $\mathrm{O}-\mathrm{O}-\mathrm{O}$ angles of $60^{\circ}-120^{\circ}$, like in the hexagonal tungsten bronze (HTB) structure.

In this three-dimensional framework, each $\mathrm{PO}_{4}$ tetrahedron shares its four apices with four $\mathrm{MO}_{6}$ octahedra and exhibits $\mathrm{P}-\mathrm{O}$ distances (Table 3) characteristic of monophosphate groups. Each $M \mathrm{O}_{6}$ octahedron is linked along b to two $\mathrm{PO}_{4}$ tetrahedra and in the $(010)$ plane to the two other $\mathrm{PO}_{4}$ tetrahedra. Thus a pecularity of the $M \mathrm{O}_{6}$ octahedra deals with the fact that they exhibit one free equatorial apex. The $M(1)$ and $M(2)$ octahedra that are about half occupied by tungsten (or molybdenum) exhibit a similar geometry (Table 3), with one shorter $M-\mathrm{O}$ bond $(1.688-1.691 \AA)$ corresponding to the free apex $(O(1)$ and $\mathrm{O}(9)$, respectively), four intermediate $M-\mathrm{O}$ distances (1.860$1.980 \AA$ ), and one abnormally long $M-\mathrm{O}$ bond (2.150$2.221 \AA$ ) corresponding to the equatorial oxygen $\mathrm{O}(6)$ or $\mathrm{O}(10)$ that is opposed the free apex. The $M(3)$ octahedra that are preferentially occupied by molybdenum are also strongly distorded. The two shortest $M(3)-\mathrm{O}$ bonds of 1.701 and $1.737 \AA$ correspond to the free oxygen atom $\mathrm{O}(15)$ and to the $\mathrm{O}(10)$ atom forming the $M(3)-\mathrm{O}(10)-M(2)$ bond, respectively, whereas the two longer bonds, $M(3)-\mathrm{O}(14)$ $(2.101 \AA)$ and $M(3)-\mathrm{O}(12)(2.181 \AA)$, are opposed to the two shortest $M(3)-\mathrm{O}$ bonds in the basal plane of these octahedra; the two intermediate $M(3)-\mathrm{O}$ distances of $1.968 \AA$ correspond to the apical $\mathrm{P}-\mathrm{O}-\mathrm{Mo}-\mathrm{O}-\mathrm{P}$ bonds along $\mathbf{b}$.

The view of this structure along (Fig. 2) shows that the whole structure can be described by the assemblage of $\left[M \mathrm{PO}_{8}\right]_{\infty}$ chains running along $\mathbf{b}$ in which one $M \mathrm{O}_{6}$


FIG. 2. The projection of the structure of $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ along $\mathbf{c}$.
octahedron alternates with one $\mathrm{PO}_{4}$ tetrahedron along $\mathbf{b}$; thus, the octahedra of one chain share one apex with the $M \mathrm{O}_{6}$ octahedra or with the $\mathrm{PO}_{4}$ tetrahedra of the adjacent chain, forming the $\left[\mathrm{MoWP}_{2} \mathrm{O}_{11}\right]_{\infty}$ framework. A threedimensional framework built up of $\left[M \mathrm{PO}_{8}\right]_{\infty}$ chains was also proposed in the structural model of $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}(5)$, but the relative orientation of the $\mathrm{PO}_{4}$ tetrahedra and octahedra within the chains and from one chain to the other are fundamentally different for the actual structure. In fact, two kinds of $\left[\mathrm{MPO}_{8}\right]_{\infty}$ chains must be distinguished according to the relative orientation of the $\mathrm{PO}_{4}$ tetrahedra with respect to each other in the same chain (Fig. 3). In the first type of chain, two successive $\mathrm{PO}_{4}$ tetrahedra exhibit a trans orientation with respect to each other (Fig. 3a) with the periodicity " $M(1)-\mathrm{P}(2)-M(1)-\mathrm{P}(3)-M(1)$." In contrast, the second kind of chain is characterized by a cis geometry of the orientation of the $\mathrm{PO}_{4}$ tetrahedra (Fig. 3b) according to the sequence " $M(1)-\mathrm{P}(1)-M(3)-\mathrm{P}(1)-M(1)$." Note that in the cis chain, one observes a pseudo-translation $01 / 20$ for all the heavy atoms and for almost all the oxygen atoms, whereas for the trans chain, only the (Mo, W) atom and some oxygen atoms respect this translation. Two identical cis chains or trans chains share the apices of their $M \mathrm{O}_{6}$ octahedra forming two sorts of $\left[M_{2} \mathrm{P}_{2} \mathrm{O}_{14}\right]_{\infty}$ double chains delimiting browmillerite-like windows, that are about $90^{\circ}$ oriented with respect to each other (Fig. 3c). Note also that a trans chain and a cis chain are linked in the following way: the $\mathrm{PO}_{4}$ tetrahedra of one chain share one apex with the $\mathrm{MO}_{6}$ octahedra of the other.

TABLE 3
Distances $(\AA)$ and Angles $\left({ }^{\circ}\right)$ in Polyhedra for $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$

| $M(1)$ | $\mathrm{O}(1)$ | $\mathrm{O}(2)$ | $\mathrm{O}(3)$ | $\mathrm{O}(4)$ | $\mathrm{O}(5)$ | $\mathrm{O}(6)$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}(1)$ | 1.688(5) | 2.671(5) | $2.696(8)$ | 2.721(7) | 2.708 (8) | 3.836(7) |
| $\mathrm{O}(2)$ | 97.2(2) | 1.868(1) | 2.742(6) | 3.824(4) | $2.753(6)$ | 2.701(5) |
| $\mathrm{O}(3)$ | 94.8(3) | 91.3(2) | 1.964(6) | 2.681(9) | 3.894(8) | 2.724(8) |
| $\mathrm{O}(4)$ | 95.5(2) | 167.2(2) | 85.7(3) | 1.980(4) | $2.739(9)$ | 2.741(6) |
| $\mathrm{O}(5)$ | 95.9(3) | 92.3(2) | 168.1(3) | 88.3(3) | 1.950(6) | 2.806 (8) |
| $\mathrm{O}(6)$ | 177.3(3) | 84.2(2) | 82.8(3) | 83.1(2) | 86.3(2) | $2.150(5)$ |
| $M(2)$ | $\mathrm{O}(7)$ | $\mathrm{O}(8)$ | $\mathrm{O}(9)$ | $\mathrm{O}(10)$ | $\mathrm{O}(11)$ | $\mathrm{O}\left(8^{\mathrm{i}}\right)$ |
| $\mathrm{O}(7)$ | 1.971(8) | $2.700(8)$ | 2.752(9) | 2.798(9) | 3.843(9) | 2.700(8) |
| $\mathrm{O}(8)$ | 87.3(2) | 1.940(6) | $2.729(8)$ | 2.757(8) | 2.739(8) | 3.847(8) |
| $\mathrm{O}(9)$ | 97.3(4) | 97.3(2) | 1.691(9) | 3.91(1) | 2.75 (1) | 2.729(8) |
| $\mathrm{O}(10)$ | 83.7(3) | 82.8(2) | 179.0(4) | $2.221(8)$ | 2.66(1) | 2.757(8) |
| $\mathrm{O}(11)$ | 163.5(4) | 90.6(2) | 99.2(4) | 79.8(3) | 1.914(9) | 2.739(8) |
| $\mathrm{O}\left(8^{\mathrm{i}}\right)$ | 87.3(2) | 165.0(3) | 97.3(2) | 82.8(2) | 90.6(2) | 1.940(6) |
| $M(3)$ | $\mathrm{O}(12)$ | $\mathrm{O}(13)$ | $\mathrm{O}(14)$ | $\mathrm{O}(15)$ | $\mathrm{O}\left(10^{\mathrm{ii}}\right)$ | $\mathrm{O}\left(13{ }^{\text {ii }}\right)$ |
| $\mathrm{O}(12)$ | 2.181(7) | $2.775(8)$ | 2.752(9) | 3.87(1) | $2.739(9)$ | 2.775(8) |
| $\mathrm{O}(13)$ | 83.8(2) | 1.968(6) | 2.690 (8) | 2.713(9) | 2.759 (8) | 3.889(8) |
| $\mathrm{O}(14)$ | 79.9(3) | 82.7(2) | 2.101(8) | 2.73(1) | 3.82(1) | 2.690(8) |
| $\mathrm{O}(15)$ | 170.9(4) | 95.1(2) | 91.0(4) | 1.701(9) | 2.65(1) | 2.713(9) |
| $\mathrm{O}\left(10^{\mathrm{ii}}\right)$ | 87.8(8) | 96.0(2) | 167.7(3) | 101.3(4) | $1.737(8)$ | 2.759(8) |
| $\mathrm{O}\left(13^{\text {iii }}\right)$ | 83.8(8) | 162.3(3) | 82.7(2) | 95.1(2) | 96.0(2) | 1.968(7) |
| $\mathrm{P}(1)$ | $\mathrm{O}(8)$ | $\mathrm{O}(13)$ | $\mathrm{O}\left(4^{\text {iv }}\right)$ | $\mathrm{O}\left(6^{v}\right)$ |  |  |
| $\mathrm{O}(8)$ | 1.554(7) | $2.446(9)$ | $2.489(9)$ | 2.483(7) |  |  |
| $\mathrm{O}(13)$ | 105.1(3) | 1.527(7) | $2.504(8)$ | 2.513(8) |  |  |
| $\mathrm{O}\left(4^{\text {iv }}\right)$ | 107.3(4) | 109.8(5) | 1.534(5) | 2.514(7) |  |  |
| $\mathrm{O}\left(6^{\mathrm{v}}\right)$ | 109.0(2) | 112.7(4) | 112.3(3) | 1.490(5) |  |  |
| $\mathrm{P}(2)$ | $\mathrm{O}(3)$ | $\mathrm{O}(14)$ | $\mathrm{O}\left(11^{\text {vi }}\right)$ | $\mathrm{O}\left(3^{\mathrm{iii}}\right)$ |  |  |
| $\mathrm{O}(3)$ | 1.534(6) | 2.537(9) | 2.51(1) | $2.420(8)$ |  |  |
| $\mathrm{O}(14)$ | 113.1(3) | 1.505(8) | 2.47(1) | 2.537(9) |  |  |
| $\mathrm{O}\left(11^{\text {vi }}\right)$ | 109.2(3) | 108.2(5) | 1.547(9) | 2.51(1) |  |  |
| $\mathrm{O}\left(3^{\text {iii }}\right)$ | 104.2(4) | 113.1(3) | 109.2(3) | 1.534(6) |  |  |
| $\mathrm{P}(3)$ | $\mathrm{O}(5)$ | $\mathrm{O}\left(7^{\text {vii }}\right)$ | $\mathrm{O}\left(12^{\text {viii }}\right)$ | $\mathrm{O}\left(5^{\mathrm{i}}\right)$ |  |  |
| $\mathrm{O}(5)$ | 1.521(6) | 2.481(9) | 2.517(8) | 2.441(8) |  |  |
| $\mathrm{O}\left(7^{\text {vii }}\right)$ | 109.0(3) | 1.529(8) | 2.463(9) | 2.481 (8) |  |  |
| $\mathrm{O}\left(12^{\text {viii }}\right)$ | 112.1(3) | 107.9(5) | 1.517(8) | 2.517(8) |  |  |
| $\mathrm{O}\left(5^{\mathrm{i}}\right)$ | 106.8(4) | 109.0(3) | 112.1(3) | 1.520(6) |  |  |

Symmetry codes

$$
\begin{array}{ll}
\text { i }: x ; 1 / 2-y ; z & \text { v }:-x ; 1-y ; 1-z \\
\text { ii : }-x ;-y ;-z & \text { vi }:-1-x ;-1-y ;-1-z \\
\text { iii: } x ; 3 / 2-y ; z & \text { vii }: x ; y ; 1+z \\
\text { iv: }-x ; y ;-z & \text { viii: }-x ;-1-y ;-1-z
\end{array}
$$

The structure of the monophosphate $\mathrm{MoWO}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ can also be described in a very simple way by considering the stacking along $\mathbf{b}$ of two kinds of layers of polyhedra labeled $A$ and $B$ (or $B^{\prime}$ ), respectively. All of them have the same formulation $\left[M_{2} \mathrm{P}_{2} \mathrm{O}_{15}\right]_{\infty}$ but exhibit a different structure.

The $A$ layers consist of disconnected infinite chains [ $\left.M_{2} \mathrm{P}_{2} \mathrm{O}_{15}\right]_{\infty}$ running along $\mathbf{c}$ (Fig. 4). In such chains the $M(1)$ bioctahedra, that exhibit the "perovskite configuration" are bridged by two $\mathrm{PO}_{4}$ tetrahedra. The $B$ (or $B^{\prime}$ ) layers are built up of the HTB bioctahedral units $M(2) M(3)$


FIG. 3. The $\left[\mathrm{MPO}_{8}\right]_{\infty}$ chains. (a) trans orientation of the $\mathrm{PO}_{4}$ tetrahedra; (b) cis orientation of the $\mathrm{PO}_{4}$ tetrahedra; (c) the assemblage of the $\left[\mathrm{MPO}_{8}\right]_{\infty}$ chains leading to the browmillerite window.
interconnected through $\mathrm{P}(2)$ and $\mathrm{P}(3)$ tetrahedra forming a bidimensional framework (Fig. 5). This $B$ layer can be described as $\left[M_{2} \mathrm{PO}_{13}\right]_{\infty}$ chains running along $\mathbf{c}$ interconnected along a through $\mathrm{PO}_{4}$ tetrahedra. In fact, two


FIG. 4. The $A$ layer $\left[\mathrm{M}_{2} \mathrm{P}_{2} \mathrm{O}_{15}\right]$.
successive $B$ layers exhibit two different orientations with respect to each other: they are turned by $180^{\circ}$ with respect to each other due to the $2_{1}$ screw axis, so that they can be named $B$ and $B^{\prime}$, respectively. Thus, the $\left[\mathrm{MoWP}_{2} \mathrm{O}_{11}\right]_{\infty}$ framework can be described by the stacking along $\mathbf{b}$ of these two kinds of layers according to the sequence " $A B A B^{\prime}$," in


FIG. 5. The $B$ layer [ $\mathrm{M}_{2} \mathrm{P}_{2} \mathrm{O}_{15}$ ].
such a way that the tetrahedra of one layer is linked to the octahedra of the next.

As a conclusion, this study has allowed a new structural type of be evidenced and shows that an important substitution of molybdenum for tungsten in the tungsten monophosphate $\mathrm{W}_{2} \mathrm{O}_{3}\left(\mathrm{PO}_{4}\right)_{2}$ can be performed without any significant modification of the structure. The green color of the crystal suggests that the octahedral sites are doped with $\mathrm{Mo}(\mathrm{V})$ in agreement with the method of growth; the existence of some oxygen vacancies is most likely but of course cannot be detected due to their very low concentration.

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